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Amidobis[η^5 -1,3-bis(trimethylsilyl)cyclopentadienyl]titanium(III)

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The title compound, $[Ti(C_{11}H_{21}Si_2)_2(NH_2)]$, crystallizes as a bent metallocene, with the NH₂ group oriented to maximize the π interaction between the N atom and the Ti^{III} center. The increased π bonding is reflected in a short Ti–N bond length of 1.933 (3) Å.

Comment

The title compound, $[Cp_2''Ti(NH_2)]$ [Cp'' is 1,3-bis(trimethylsilyl)cyclopentadienyl], (I), crystallizes in the primitive mono $clinic space group <math>P2_1/n$ with four discrete molecules in the unit cell. Selected distances and angles are given in Table 1. Fig. 1 shows the molecular geometry and the atom-numbering scheme.



The Ti center is coordinated by two virtually planar cyclopentadienyl rings, in a staggered conformation relative to each other, and by an NH₂ group. The angle formed by the cyclopentadienyl ligands, Cg1-Ti-Cg2, is 137.90 (8)° (Cg1 and Cg2 are the centroids of the cyclopentadienyl rings C1-C5 and C6-C10, respectively). The distances between the Ti atom and Cg1 and Cg2 are 2.0611 (17) and 2.0623 (17) Å, respectively. These are comparable with values found for other Cp₂TiNR₂ derivatives (Brady *et al.*, 1995; Lukens *et al.*, 1996; Feldman & Calabrese, 1991).



Figure 1

A view of the structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.

The amide group is located within an open wedge formed by the sterically demanding trimethylsilyl groups, allowing the least steric bulk and a proper orientation for the filled *p* orbital to interact with the empty b_2 orbital of the Cp₂''Ti fragment, enabling the NH₂ moiety to act as a π donor. This is also reflected in the small Cg-Ti-N-H torsion angle of 6.4° and the quite short Ti-N distance of 1.933 (3) Å. Similar N- $M\pi$ bonding is also observed in $[(\eta^5-C_5Me_5)_2Ti(NH_2)]$ [Ti-N = 1.944 (2) Å; Brady *et al.*, 1995), $[(\eta^5-C_5Me_5)_2Ti(NHMe)]$ [Ti-N = 1.955 (2) Å; Lukens *et al.*, 1996] and $[(\eta^5-C_5Me_5)_2-Hf(H)(NHMe)]$ [Hf-N = 2.027 (8) Å; Hillhouse *et al.*, 1988], as evidenced by a similar orientation of the amide groups as well as short (Ti,Hf)-N distances.

The influence of steric bulk *versus* electronic effects is demonstrated in bis(pentamethylcyclopentadienyl)titanium(III) *N*-methyl-*N*-phenylamide (Feldman & Calabrese, 1991), in which the amide ligand adopts a perpendicular orientation with respect to the plane formed by the Ti atom and the two cyclopentadienyl ring centroids, preventing the *p* orbital from interacting with the b_2 orbital of the $[(\eta^5-$ C₅Me₅)₂Ti] fragment. Therefore, the Ti—N distance [2.054 (2) Å] is elongated compared with that of (I).

Experimental

The title compound was synthesized from Cp_2'' TiMe and dry NH₃ in *n*-hexane in a manner similar to that used for the preparation of $(C_5Me_5)_2Ti(NH_2)$ (Brady *et al.*, 1995). Dichroic (green and brown) crystals were grown by slow cooling of a saturated hexane solution to 193 K. The crystal was handled under a nitrogen atmosphere, mounted on a glass fiber with Paratone-N hydrocarbon oil and cooled to 133 (2) K for data collection.

metal-organic compounds

 $D_x = 1.128 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation

reflections

 $\mu = 0.48~\mathrm{mm}^{-1}$

T = 133 (2) K

 $R_{\text{int}} = 0.059$ $\theta_{\text{max}} = 26.1^{\circ}$

 $h = -16 \rightarrow 13$

 $\begin{array}{l} k=-20\rightarrow 18\\ l=-15\rightarrow 14 \end{array}$

 $\theta = 2-51.2^{\circ}$

Cell parameters from 5983

Block, dichroic (green-brown)

5071 independent reflections

3874 reflections with $I > 2\sigma(I)$

 $0.38 \times 0.32 \times 0.18 \text{ mm}$

Crystal data

$[Ti(C_{11}H_{21}Si_2)_2(NH_2)]$	
$M_r = 482.84$	
Monoclinic, $P2_1/n$	
$a = 13.0945 (3) \text{\AA}$	
b = 16.5699 (3) Å	
c = 13.1869 (2) Å	
$\beta = 96.488 (1)^{\circ}$	
V = 2842.90 (9) Å ³	
Z = 4	
Data collection	
Bruker SMART 1K CCD area-	
detector diffractometer	
w scaps	
w scans	
Absorption correction: multi-scan	

Absorption correction: multi-scan (Blessing, 1995) $T_{\min} = 0.839, T_{\max} = 0.919$ 13 362 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.113$ S = 1.13 5071 reflections 273 parameters H atoms treated by a mixture of independent and constrained	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0201P)^{2} + 5.0672P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.47 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.32 \text{ e } \text{\AA}^{-3}$
independent and constrained refinement	

Table 1

Selected geometric parameters (Å, °).

Cg1 is the centroid of the C1–C5 ring and Cg2 is the centroid of the C6–C10 ring.

Ti1-C1	2.426 (3)	Ti1-C8	2.366 (3)
Ti1-C2	2.376 (3)	Ti1-C9	2.382 (3)
Ti1-C3	2.377 (3)	Ti1-C10	2.399 (3)
Ti1-C4	2.364 (3)	Ti1-N1	1.933 (3)
Ti1-C5	2.393 (3)	Ti1 - Cg1	2.0611 (17)
Ti1-C6	2.430 (3)	Ti1 - Cg2	2.0623 (17)
Ti1-C7	2.372 (3)	-	
Ti1-N1-H23A	128 (3)	Cg1-Ti1-Cg2	137.90 (8)
Ti1-N1-H23B	127 (3)	Cg1 - Ti1 - N1	111.13 (12)
H23A-N1-H23B	105 (4)	cg2-Ti1-N1	110.97 (12)

Methyl H atoms of the trimethylsilyl groups, as well as H atoms of the cyclopentadienyl rings, were included in calculated positions, with C–H distances of 0.98 (CH₃) and 0.95 Å (CH), and were refined using a riding model, with $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})$ for the methyl H atoms and $1.2U_{\rm eq}({\rm C})$ for the cyclopentadienyl H atoms. The amide H atoms (H23A and H23B) were located using standard difference Fourier techniques and were refined isotropically; the N–H distances are 0.85 (4) and 0.87 (5) Å.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEP*-3 (Farrugia, 1997); software used to prepare material for publication: *SHELXL*97.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FA1076). Services for accessing these data are described at the back of the journal.

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